

Fig. 2. A projection of the structure of WO_2 on to the (010) plane.

such as VO₂, TcO₂ and NbO₂ (Rogers, Shannon, Sleight & Gillson, 1969) as well as in AlWO₄ (Doumerc, Vlasse, Pouchard & Hagenmuller, 1975) and GaWO₄ (Palmer, 1977).

There is no evidence for the space group $P2_1$ as suggested by Magnéli & Andersson (1955) from either the X-ray or neutron powder profiles: the symmetry and size of the unit cell of WO₂ are such that there would be no fully resolved peaks h0l, l = 2n + 1 in the 2θ range investigated. However, refinement was attempted in $P2_1$; no improvement in the agreement between observed and calculated profiles was seen, and indeed the atomic coordinates refined effectively to the symmetry-equivalent positions in $P2_1/c$.

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Dieuropium(III) Germanate Oxide

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Abstract. Eu₂(GeO₄)O, monoclinic, $P2_1/c$, a = 9.977 (7), b = 7.136 (4), c = 6.854 (4) Å, $\beta = 115.78$ (6)°, Z = 8, $D_m = 6.84$ (2), $D_x = 6.90$ Mg m⁻³. Crystals were prepared by melting a pre-sintered mixture of Eu₂O₃ and GeO₂ in an image furnace with a tungsten halogen lamp as heat source. The seven-coordination polyhedra of Eu³⁺ ions in a set of

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equivalent sites form slabs parallel to (100) by edgesharing. These slabs are connected to the corrugated layers of the edge-shared polyhedra of sevencoordinated Eu^{3+} ions in other equivalent positions through common edges of the polyhedra and O–Ge–O bonds of the GeO₄ tetrahedra.

Introduction. Of the germanates and silicates of the rare-earth elements Ln_2XO_5 (Ln = Sc, Y, La, Ce, ..., © 1979 International Union of Crystallography

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Lu; X = Si, Ge), the crystal structures have been determined for Sc₂GeO₅ (Gorbunov, Maksimov, Kharitonov & Belov, 1974) and Y₂SiO₅ (Michel, Buisson & Bertaut, 1967). The germanates of Lu, Yb, Er, Ho, Dy, Tb and Y are isomorphous with Y₂SiO₅ (Buisson & Michel, 1968). Leonov & Bondar (1973) classified the compounds Ln₂GeO₆ into two subgroups of different structure types and stated that the germanates, Ln₂GeO₅, of La, Ce, Pr, Nd, Sm and Eu with larger ionic radii than Gd³⁺ are orthorhombic while those of Tb, Dy, Ho, Y, Er, Tm, Yb, Lu and Sc with smaller ionic radii are monoclinic. The crystal structure of Eu₂GeO, was investigated because the crystals prepared by the present authors were found to be monoclinic and to adopt an unknown structure which is evidently not isotypic with those of Sc₂GeO₅ and Y₂SiO₄.

The lattice constants were refined by least squares from the 2θ values of 19 reflections measured on a four-circle diffractometer (Rigaku) with Mo Ka radiation $[\lambda(Mo K\alpha_1) = 0.70926 \text{ Å}]$ monochromatized with a pyrolytic-graphite crystal ($2\theta_M =$ 12.17°). The systematic absences h0l with $l \neq 0 \pmod{l}$ 2) and 0k0 with $k \neq 0 \pmod{2}$ indicated space group $P2_1/c$. Within the limits of $2\theta \leq 90^\circ$, $h \ge 0$ and $k \ge 0$, the intensities of 3782 reflections were collected on the same diffractometer operated in the 2θ - ω -scanning (when $2\theta \ge 30^{\circ}$) or in the ω -scanning mode ($2\theta < 30^{\circ}$). Those reflections whose net intensities (I_{o}) did not exceed the e.s.d.'s (σ_{i}) were considered to be unobserved. The 2806 observed reflections were corrected for absorption [numerical integration with $8 \times 8 \times 4$ Gaussian points in the specimen bounded by six faces; specimen size $0.10 \times$ $0.07 \times 0.02 \text{ mm}, \mu(\text{Mo } K\alpha) = 33.5 \text{ mm}^{-1}$ and then reduced to a set of 2706 independent reflections.

The positions of the Eu and Ge atoms were determined from the Patterson function by means of a minimum function (Buerger, 1959). The O atoms were located by a difference synthesis. The least-squares refinement of the coordinates and anisotropic thermal parameters of the atoms was performed with a program written by one of the authors. The sum of $w(|F_o| - |F_c|)^2$ with $w = 1/\sigma^2$ and $\sigma = \sigma_1 F_o/(2I_o)$ was minimized. The scattering factors given by Cromer & Mann (1968) were used with corrections for dispersion (Cromer & Liberman, 1970). A correction was also made for extinction with the formula of Zachariasen (1968) in the following simplified form, modified for the partially polarized incident beam:

$$F_c = sF[1 + (g/\sin 2\theta)(1 + \cos^4 2\theta \cos^4 2\theta_M) \times (1 + \cos^2 2\theta \cos^2 2\theta_M)^{-1} (sF)^2]^{-1/4},$$

where $g = 2.7 (1) \times 10^{-6}$, s = 1.267 (4). The coefficient g was treated as one of the free parameters to be refined; s is a scale factor. The atomic coordinates

and isotropic thermal parameters calculated from the anisotropic ones are listed in Table 1.* $R = \sum ||F_o| - |F_c|| / \sum |F_o| = 0.050$ for the 2706 non-zero reflections. The parameter shifts in the last cycle were <0.6% of the e.s.d.'s. The last difference syntheses with $\Delta F = F'(|F_o| - |F_c|)/|F_c|$, where F_c and F' represent the structure factors calculated with and without the extinction and dispersion corrections respectively, revealed no further significant details of the structure, except that ripples of electron density within the limits of $\pm 8 \text{ e} \text{ Å}^{-3}$

Table 1. Atomic coordinates and isotropic thermal parameters of Eu₂GeO₅

	x	У	Z	B (Å ²)
Eu(1)	0.13006 (6)	0.13492 (9)	0.18212 (8)	0.38(1)
Eu(2)	0.47725 (6)	0.12804 (9)	0.70949 (8)	0.39(1)
Ge	0.80231 (14)	0.08932 (18)	0.27673 (18)	0.37(2)
O(1)	0.1086 (11)	0.0543 (15)	0.8315 (14)	0.81 (17)
O(2)	0-2146 (9)	0.0790 (14)	0.5483 (13)	0.60 (19)
O(3)	0.3904 (9)	0.1176 (14)	0.3218 (12)	0.52 (22)
O(4)	0.6107 (10)	0.1347 (16)	0.1166 (13)	0.59 (37)
O(5)	0.8894 (11)	0.2092 (13)	0.8931 (15)	0.58 (38)



Fig. 1. Projection of the Eu_2GeO_5 structure on to (010). Numbers indicate the heights of atoms in hundredths of the *b* axis.



Fig. 2. Structure of Eu₂GeO₅ viewed along c.

^{*} Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34519 (17 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

were found around both Eu positions. These ripples might be due to the small systematic errors introduced by the extinction and/or the absorption correction(s).

Discussion. Figs. 1 and 2 show the Eu₂GeO₅ structure projected parallel to [010] and [001] respectively. While the small Sc^{3+} ions $[r(^{v_1}Sc^{3+}) = 0.745 \text{ Å};$ Shannon & Prewitt (1970)] in Sc₂GeO, are each surrounded octahedrally by six O atoms (Gorbunov et al., 1974), there are two crystallographically independent Y^{3+} ions $[r(^{VI}Y^{3+}) = 0.900 \text{ Å}]$ in Y_2SiO_3 , one of which is coordinated to six and the other to seven O atoms (Michel et al., 1967). In Eu₂GeO₅, both Eu(1) and Eu(2) have seven O-atom nearest neighbours in accordance with their larger ionic radii $[r(^{VI}Eu^{3+})] =$ 0.947, $r(^{VII}Eu^{3+}) = 1.03$ Å]. The coordination polyhedra [monocapped octahedra (Wells, 1975)] of Eu(2) are joined by edge-sharing to form a slab with quasiclose-packed O-atom layers parallel to (100) (Fig. 3). The coordination around Eu(1) can be regarded as a distorted monocapped trigonal prism. The coordination polyhedra of Eu(1) are connected to each other through common edges and form a corrugated layer parallel to (100) together with the equivalent polyhedra in the neighbouring unit cells (Fig. 4). The slabs



Fig. 3. Slab of the edge-shared coordination polyhedra of Eu(2) viewed along a.



Fig. 4. Corrugated layer of the edge-shared coordination polyhedra of Eu(1) and the GeO_4 tetrahedra viewed along **a**.

Table 2. Eu–O distances (Å)

E.s.d.'s are in parentheses.

Symme	etry code	;				
(i)	1 - x,	1 - y,	1 - z	(v)	х,	y, 1 + z
(ii)	х,	$\frac{1}{2} - y$,	$\frac{1}{2} + z$	(vi)	-1 + x,	y, -1 + z
(iii)	х,	$\frac{1}{2} - y$, -	$\frac{1}{2} + z$	(vii)	х,	y, -1 + z
(iv)	1 - x,	-y,	1 - z			
Eu(1)-	-O(1 ^{vii})	2.389	(9)	Eu(2)-	-O(2)	2.386 (9)
	O(1 ⁱⁱⁱ)	2.491	(10)		O(3)	2.412(7)
	O(2)	2.308	(8)		O(3 ^{iv})	2.260(10)
	$O(2^{iii})$	2.529	(9)		O(3 ¹¹)	2.284(10)
	O(3)	2.349	(9)		O(4 ^{iv})	2.572 (10)
	O(5 ^{iv})	2.500	(10)		O(4 ¹¹)	2.404(10)
	O(5 ^{vl})	2.415	(10)		O(4 ^v)	2.516 (8)
Mean		2.426		Mean		2.404

Table 3. Bond lengths (Å) and angles (°) around the Ge atom

Ge-O distances are given on the diagonal; O-O distances and O-Ge-O angles in the upper-right and the lower-left triangles respectively. E.s.d.'s are in parentheses. The symmetry code is given in Table 2.

	O(1 ^{iv})	O(2 ^{iv})	O(4)	O(5 ⁱⁱⁱ)
O(1 ^{iv})	1.721 (9)	2.59(1)	2.99 (1)	2.91 (1)
O(2 ^{iv})	96.2 (4)	1.757 (9)	2.68(1)	2.93 (1)
O(4)	117.5 (4)	98.6 (4)	1.772 (9)	2.82(1)
O(5 ¹¹¹)	117.0 (5)	116.2 (4)	109.4 (5)	1.690 (9)
Mean Ge	e–O 1.735			

of Eu(2) and the corrugated layers of Eu(1) are linked through a common edge O(2)-O(3) and further by O-Ge-O bonding of the GeO₄ tetrahedra.

It can be seen from Figs. 1 or 2 that there exist only four Eu-O bonds across the unit-cell boundary parallel to (100), suggesting the relative weakness of the bonding in this direction as compared with the other directions. Moreover, the electrical neutrality can always be maintained when the crystal is divided into two parts at any unit-cell boundary. These facts might well explain the perfect cleavage of the crystal parallel to (100). Prior to the structure determination, measurements of emission and absorption spectra of Eu₂GeO, had been carried out by one of the authors. The results had predicted that the Eu³⁺ ions occupy two sets of equivalent positions with low point symmetry. The bond lengths and angles were calculated by means of a local version of the ORFFE program (Busing, Martin & Levy, 1964), and are tabulated in Tables 2 and 3.

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